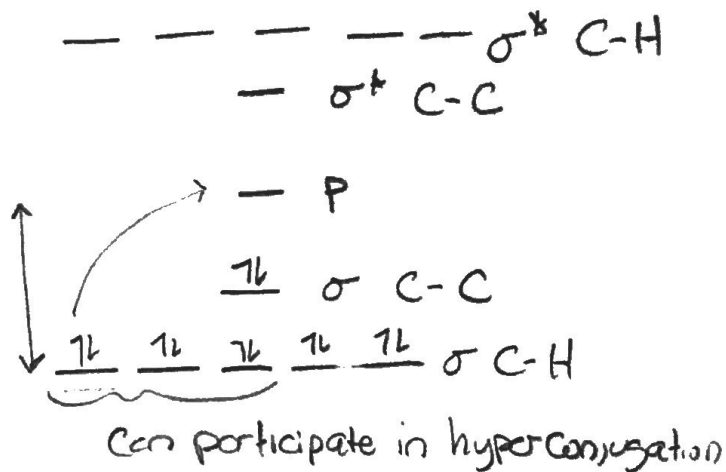
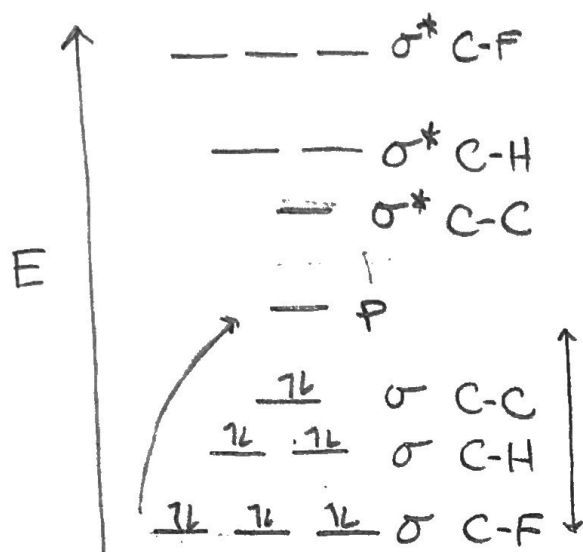
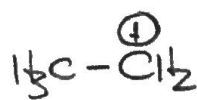
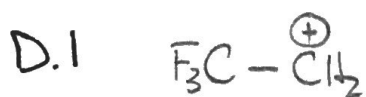


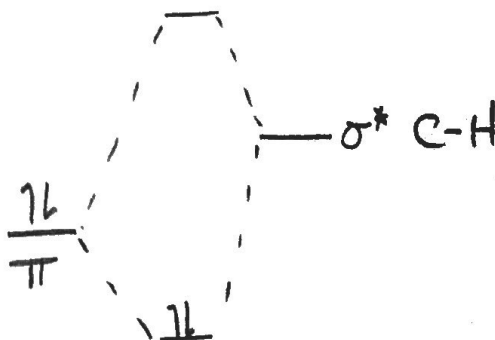
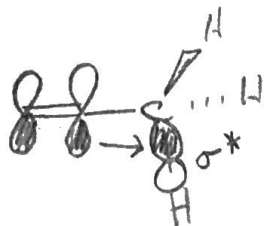
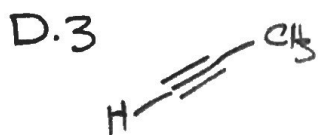
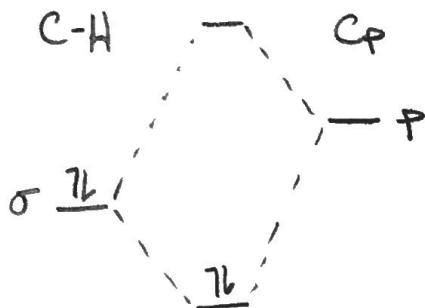
Intro to MO Theory - HW Solns



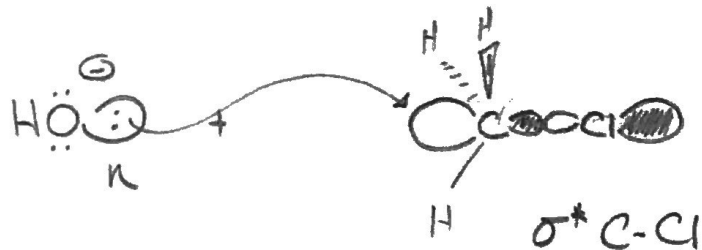
Smaller energy gap = better interaction

The hyperconjugation is better in $H_3C-CH_2^+$ than in $F_3C-CH_2^+$

MO Interaction:

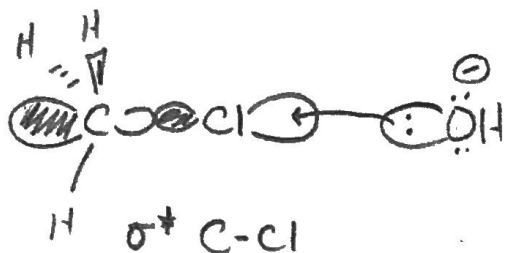


D.5



Correct approach
to σ^* at the C

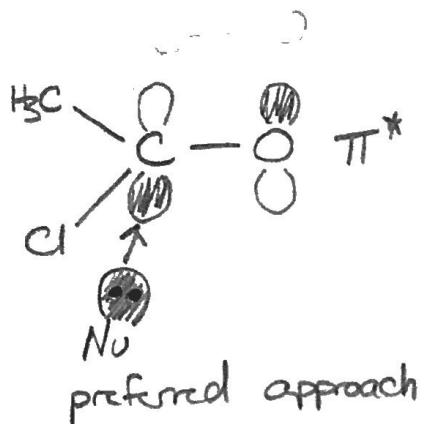
It is allowed (constructive orbital overlap) for HO^- to approach σ^* from the Cl side



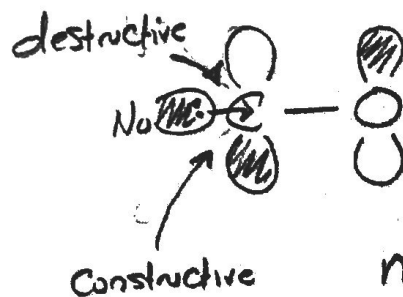
This does not happen, however,
due to electrostatics.

This is a site of high e^- density
reacting with another site of
high e^- density, which is not
favorable.

D.7 Nu will react with the LUMO, which is the π^* orbital

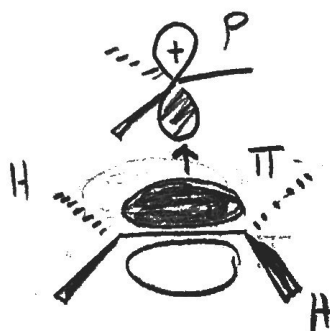
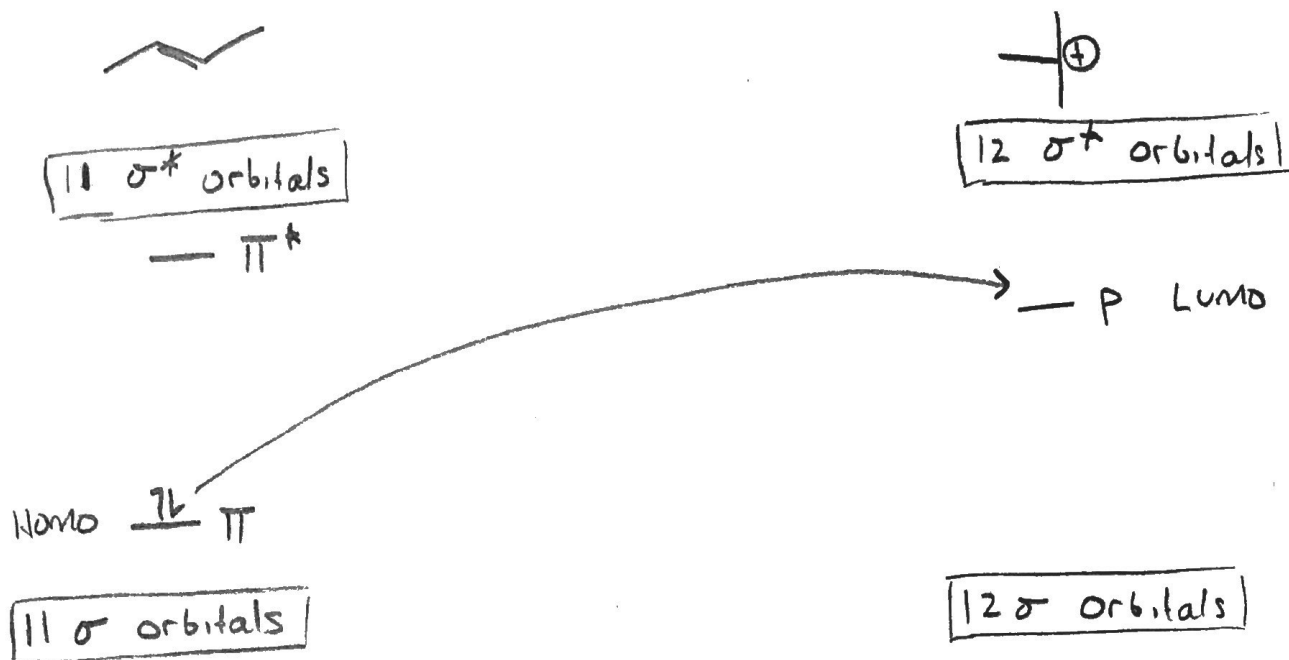


If nu approaches directly along
the C=O bond



no net gain
= does not occur

D.8



Constructive
interaction